

## Stechiometria

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### Stechiometria

Stoichiometry / , s t o i k i ' b m i t r i / is the calculation of reactants and products in chemical reactions in chemistry.. Stoichiometry is founded on the law of conservation of mass where the total mass of the reactants equals the total mass of the products, leading to the insight that the relations among quantities of reactants and products typically form a ratio of positive integers.

### Stoichiometry - Wikipedia

stoi·chi·om·e·try (stoi'kē-ōm'ī-trē) n. 1. Calculation of the quantities of reactants and products in a chemical reaction. 2. The quantitative relationship between reactants and products in a chemical reaction. [Greek stoikheion, element; see steigh- in Indo-European roots + -metry.] stoi'chi·o·met'ric (-ə-mēt'rik) adj. stoi'chi ...

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### English Translation of "stechiometria" | Collins Italian ...

How to Do Stoichiometry. In a chemical reaction, matter can neither be created nor destroyed according to the law of conservation of mass, so the products that come out of a reaction must equal the reactants that go into a reaction. This...

### How to Do Stoichiometry (with Pictures) - wikiHow

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### Stechiometria - Wikipedia

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(uncountable, chemistry) The study and calculation of quantitative (measurable) relationships of the reactants and products in chemical reactions (chemical equations).· (countable, chemistry) The quantitative relationship between the reactants and products of a specific reaction or equation

### stoichiometry - Wiktionary

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