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Title: 2014 HKDSE Chem Examination - C - Lv1 Additional Samples Author: HKEAA Created Date: 20140827180714Z

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- 8 - (ii) \dot{U} First, Ag_2CrO_4 (s) dissociates. \dot{U} At a particular instance, the rate of the forward and backward reactions becomes equal to each other. Ag_2CrO_4 (s) \rightleftharpoons 2Ag^+ (aq) + CrO_4^{2-} (aq) \dot{U} From that instant, concentration of ions remain constant, the ionic product, K_{sp} of the compound at the given temperature.

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